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operate in the presence of  
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catalyzes only one reaction,

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there is almost always a  
change in energy, which can  
be observed as a change in  
temperature of the reaction  
mixture. Exothermic  
reactions release energy,  
resulting in an increase in  
temperature. Endothermic  
reactions absorb energy,  
resulting in a decrease in  
temperature.

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Answers Chemical Change Study  
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15-2 Chemical Energy and the  
Universe (cont.) • Chemists  
are interested in changes in  
energy during reactions. •  
Enthalpy is the heat content  
of a system at constant  
pressure. • Enthalpy (heat)  
of reaction is the change in  
enthalpy during a reaction  
symbolized ...

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triphosphate (ATP) 2.a  
molecule that transfers  
energy from the breakdown of  
food molecules to cell  
processes 3. ATP is a high-  
energy molecule that is  
converted into lower-energy  
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Absorption of light= Primary pigment chlorophyll. 2. Conversion of light energy to chemical energy. 3. Storage of energy and organic molecules= Calvin Cycle. Which stages would be considered a light reaction. Light reactants. Which stages would

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### Answer

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In your textbook, read about cell energy. Use each of the terms below just once to complete the passage. energy phosphate adenine charged

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ATP chemical bonds work  
ribose To do biological (1)  
\_\_\_\_\_, cells require  
energy.

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